

# Development of AMPYRE - A Dynamic Pyroprocess Model

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U. S. DEPARTMENT OF  
**ENERGY**

**Nuclear Energy**

# Development of Argonne Model for PYrochemical Recycling (AMPYRE)

- Mass-balance model of a complete electrochemical facility
  - User interface, calculations, and formatted output all programmed using Microsoft Excel with Excel VBA
- Unit operations of facility, with interdependency
  - Head end operations
  - U and U/TRU product collection
  - Salt recovery, treatment, and recycle
  - Wasteform fabrication
  - Predicts batch-wise evolution of salt and product compositions
  - Simple variables representing chemical behavior
    - Chemistry-based models being developed for anode and cathode subunits
- Can be used to explore
  - Effect of changing operating conditions or fuel composition
  - Impact of variations in inputs
  - Flowsheet option comparisons

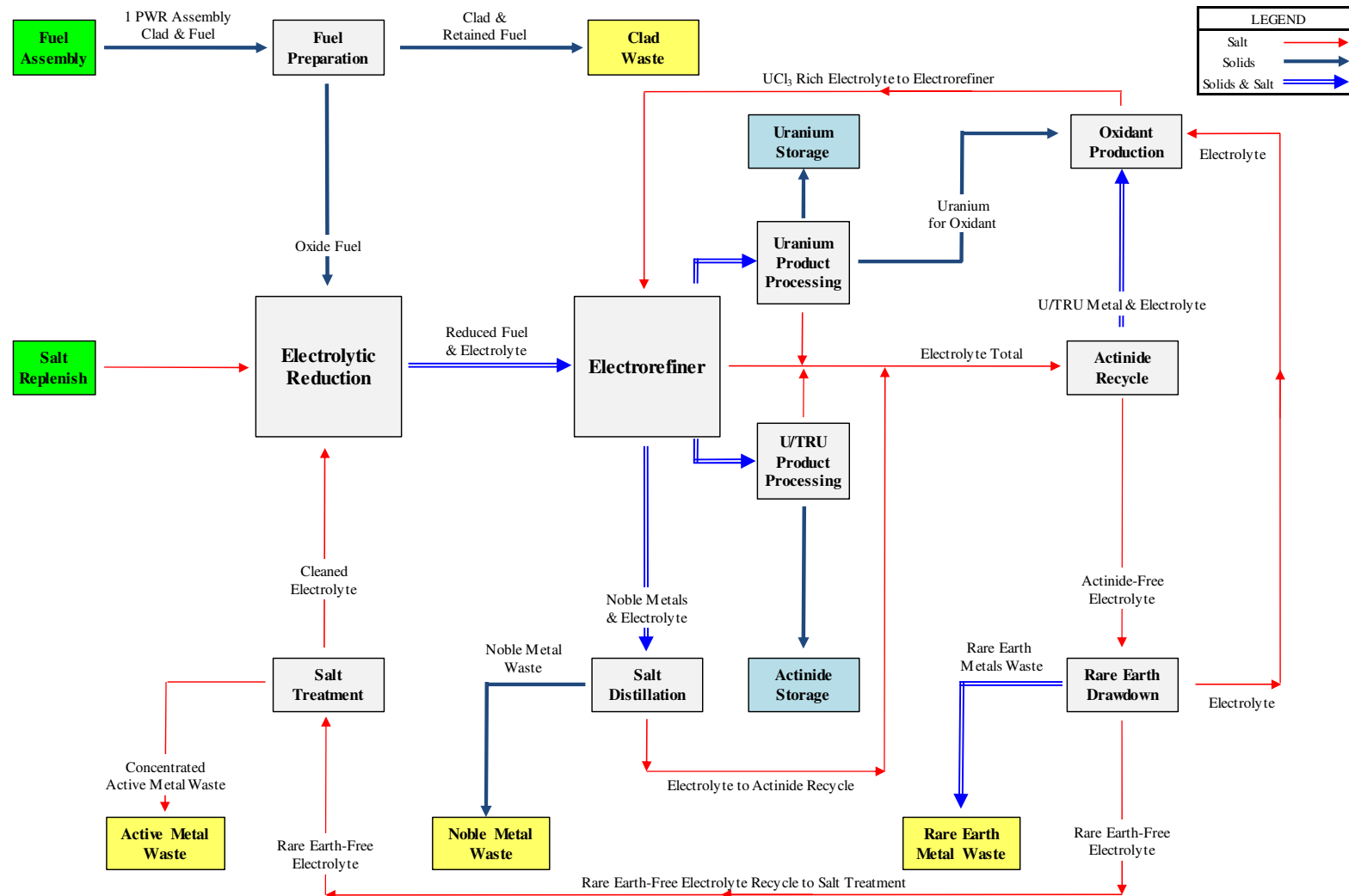


# Key Features of A Fully-Integrated, Dynamic Pyrochemical Process Model

- Accurately models all separations steps in the process
  - Chemical separation & conversion steps (e.g. oxide reduction, electrorefining, actinide recovery, oxidant production, etc.)
  - Physical separation steps (U-product processing, U/TRU product processing)
  - Wasteform production (metal and ceramic wasteforms)
- Accurately tracks evolution of salt and product compositions as material moves through an electrochemical reprocessing facility
  - Material transfers of discrete batches of solids with solid adhered salt
  - Material transfers of molten salt
  - Batch and semi-continuous process operations
- Accurately handles transition from start-up to steady state
  - Not all process operations are performed initially (e.g. U/TRU deposition, actinide drawdown, rare earth drawdown, wasteform production)
  - Processes “turn on” when concentrations of key species in salt build up to target levels



# Flowsheet for LWR Fuel



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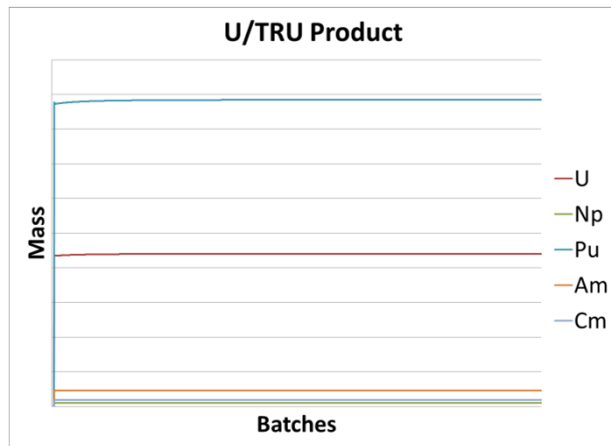
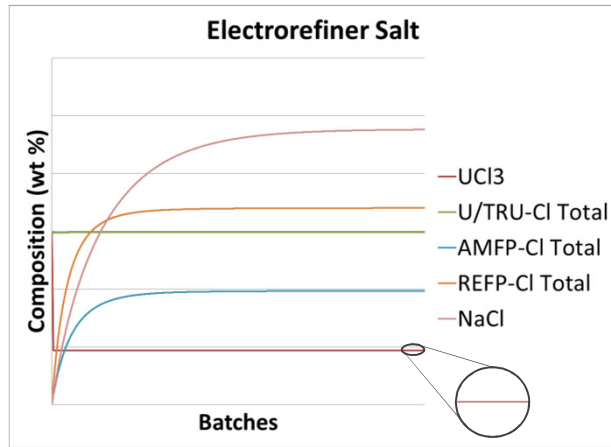
# A User-Friendly Dynamic Model

- Model was built using MS Excel and VBA
- Automated handling of operating changes and resumption of existing runs
- Revised code to handle variable inputs
- Import user-defined fuel composition from spreadsheet
- Expanded options for Cs/Sr recovery

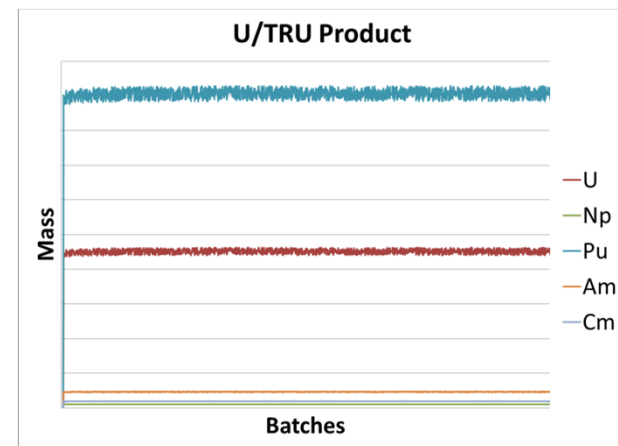
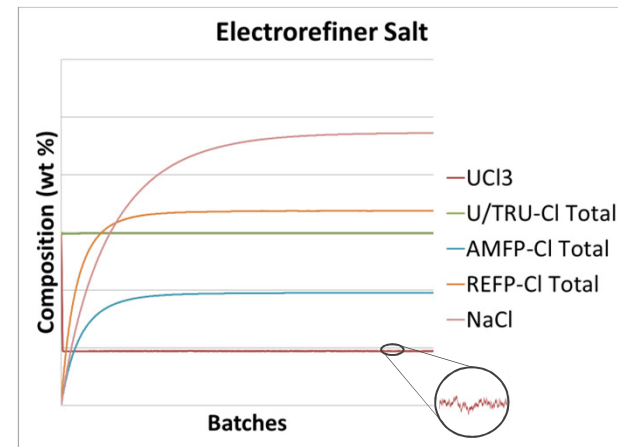
The image displays three overlapping screenshots of a software interface for a dynamic model, likely for nuclear fuel cycle simulation. The top window shows the 'General' and 'System Parameters' tabs, with fields for 'Name for results file', 'Directory (folder) name', 'Starting Iteration', and 'Final Iteration'. The middle window shows the 'Recovery and Transfer Parameters' tab, featuring a table for 'Activation Products (cont'd)' and 'Fission Products (cont'd)'. The bottom window shows the 'Electrorefiner Salt' tab, which includes sections for 'Electrolyte', 'Rare Earths', 'Active Metals', 'Actinides', and 'Electrotransport and Product Requirements'.

# Sensitivity Analysis: Variation in Metal Fuel Composition Input

Constant Fuel Composition

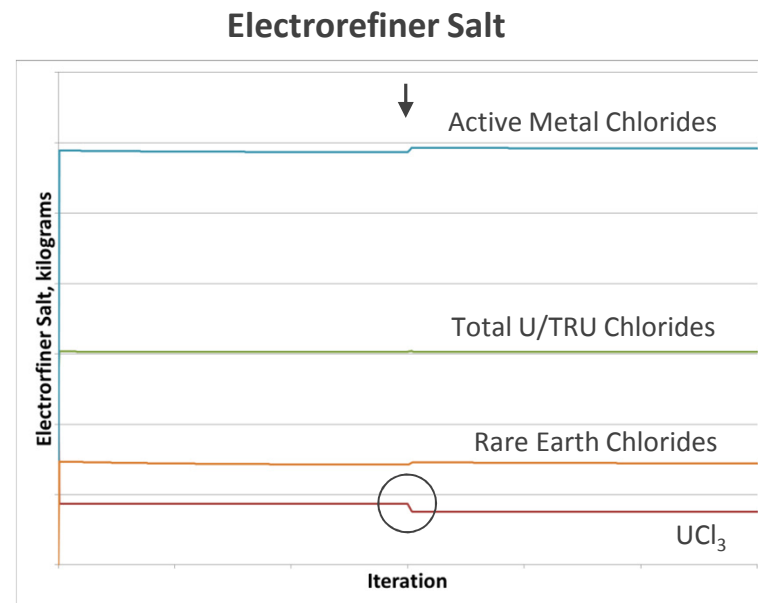
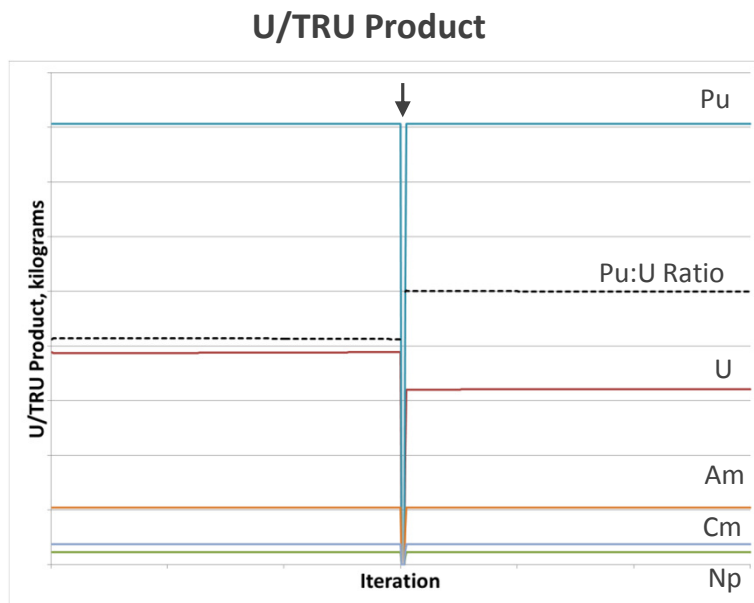


Variation in Pu Content of Fuel



# System Response to Changes

- Ability to adjust to operational changes



- User interface can load previous results and resume calculation
- Predicting effect of operational changes is key to materials tracking



# Development of Dynamic Electrorefiner Model (DyER)

- Time-dependent, current-based electrorefiner model
- Calculations in MATLAB<sup>®</sup>
- Electrochemical models of anode and cathode behavior
- Simplified approach to salt recycle
  - To be used in combination with mass-balance facility model
- Predicts time-dependent evolution of salt and product compositions from user-defined operating conditions:
  - Initial salt composition and fuel composition
  - Number of cathodes; anode and cathode surface area
  - Operation at constant voltage or constant current
  - Adjustable oxidant addition
  - Can be used to explore transient conditions in the electrorefiner
- Can be interfaced with AMPYRE



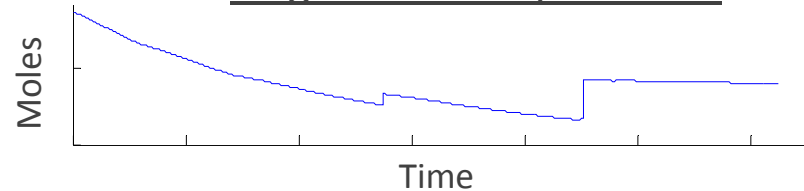
# Example of Dynamic Electrefiner (MATLAB)

## Model Results

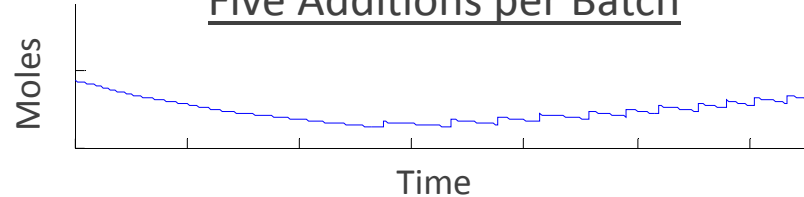
- Effect of Method of Oxidant Addition
  - Processing multiple batches
  - Varying frequency of oxidant addition
    - Single addition per batch
    - Multiple additions per batch
    - Continuous addition of oxidant
  - Scale of observable response

$\text{UCl}_3$  in Electrefiner Salt

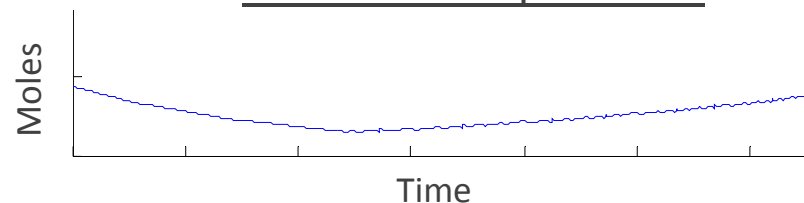
Single Addition per Batch



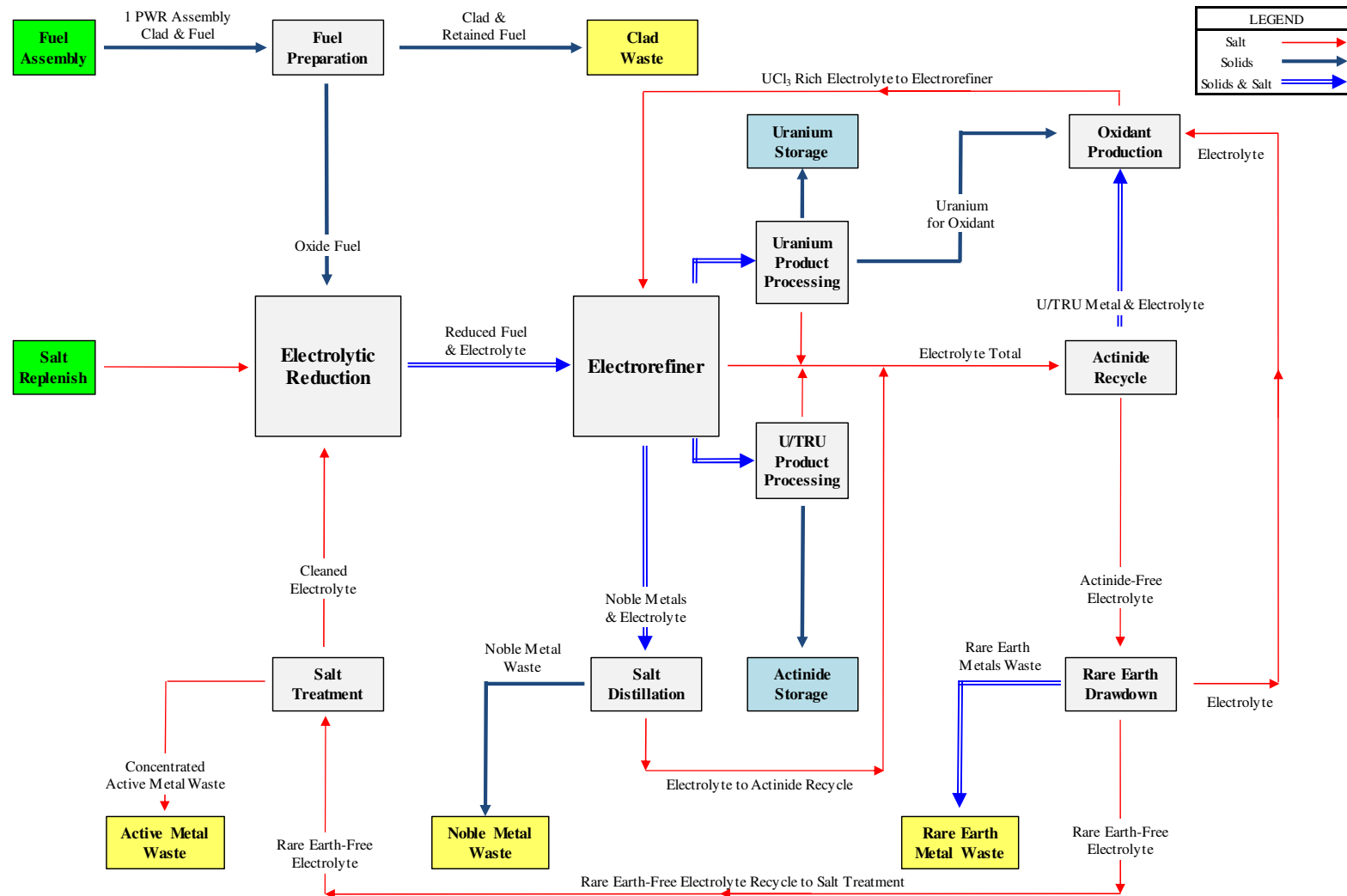
Five Additions per Batch



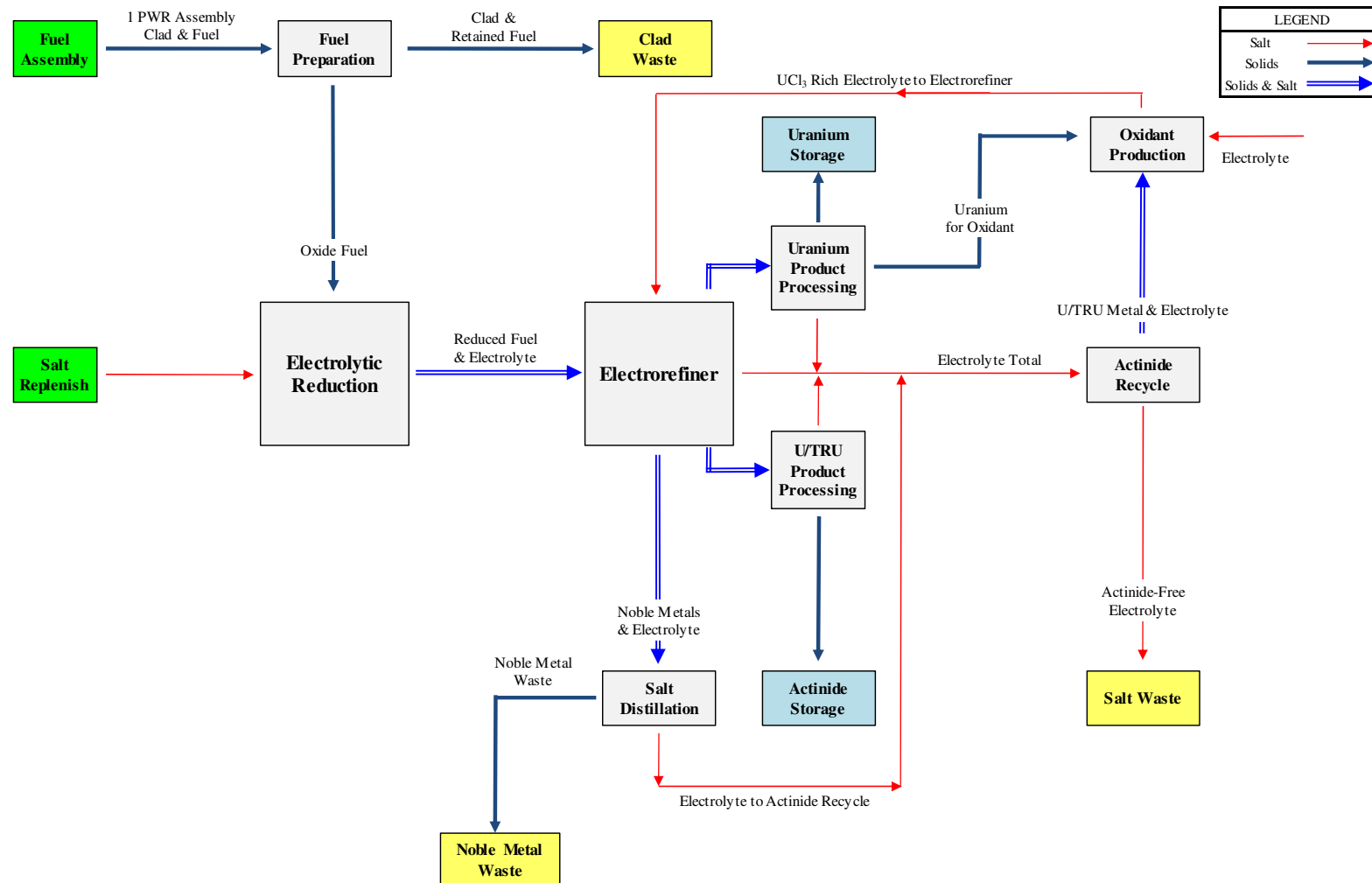
Ten Additions per Batch



# Flowsheet with Salt Recycle

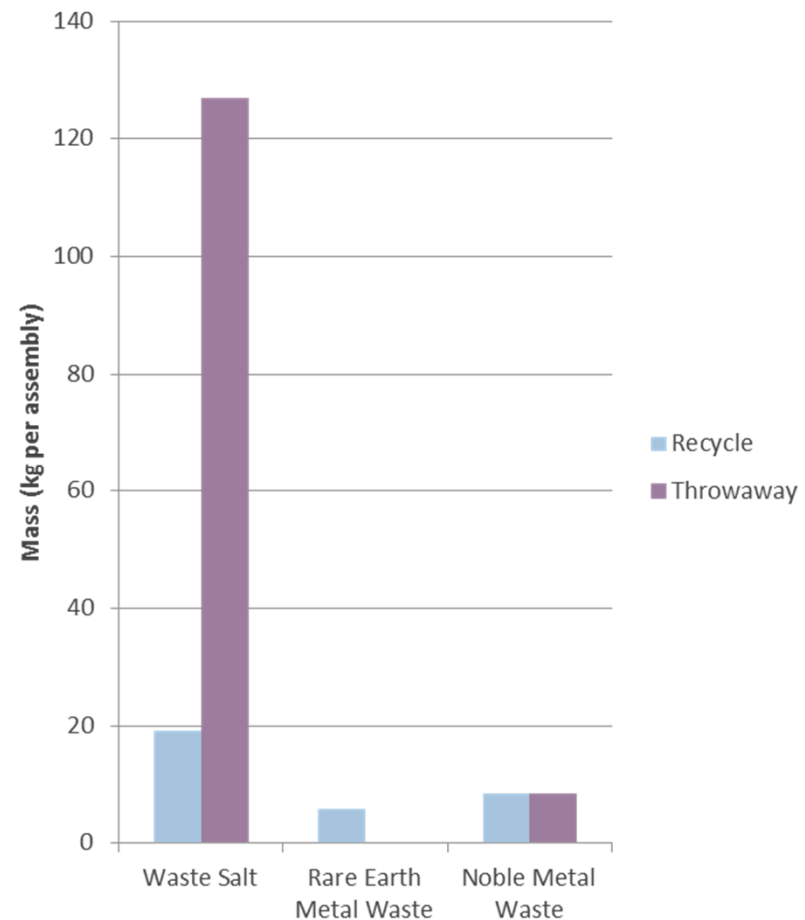


# Flowsheet with Direct Throwaway



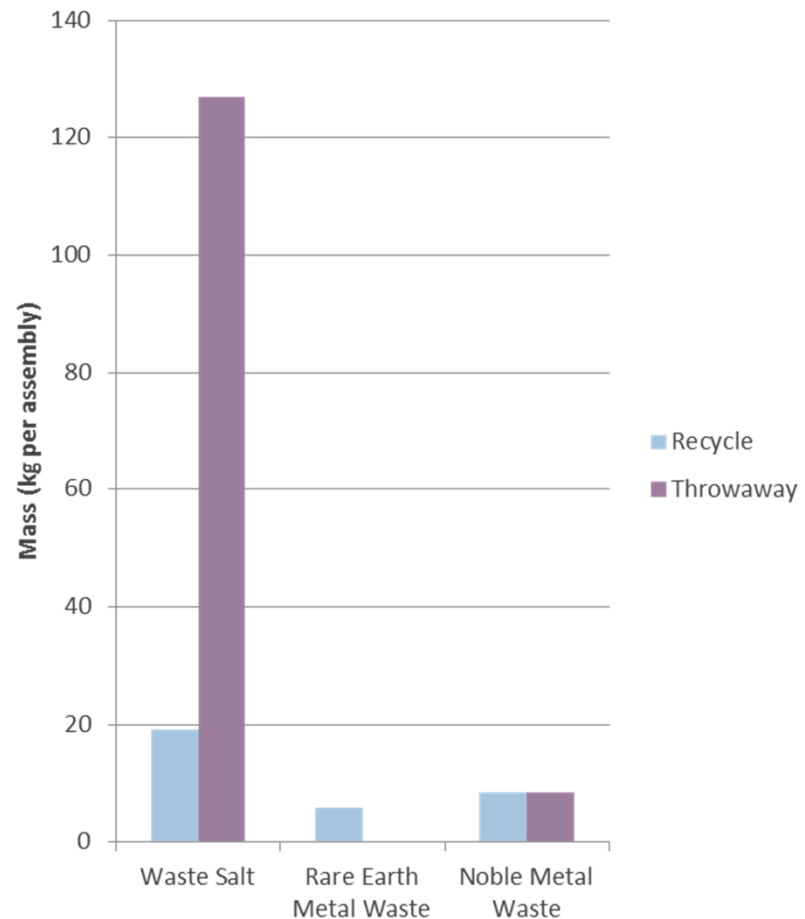
# Using AMPYRE to Evaluate Flowsheet Options (Oxide Fuel)

- Comparison of salt handling options
  - Recycle loop to remove rare earth and active metal fission products
  - Direct throwaway of electrorefiner salt following recovery of U/TRU
- Impact on waste

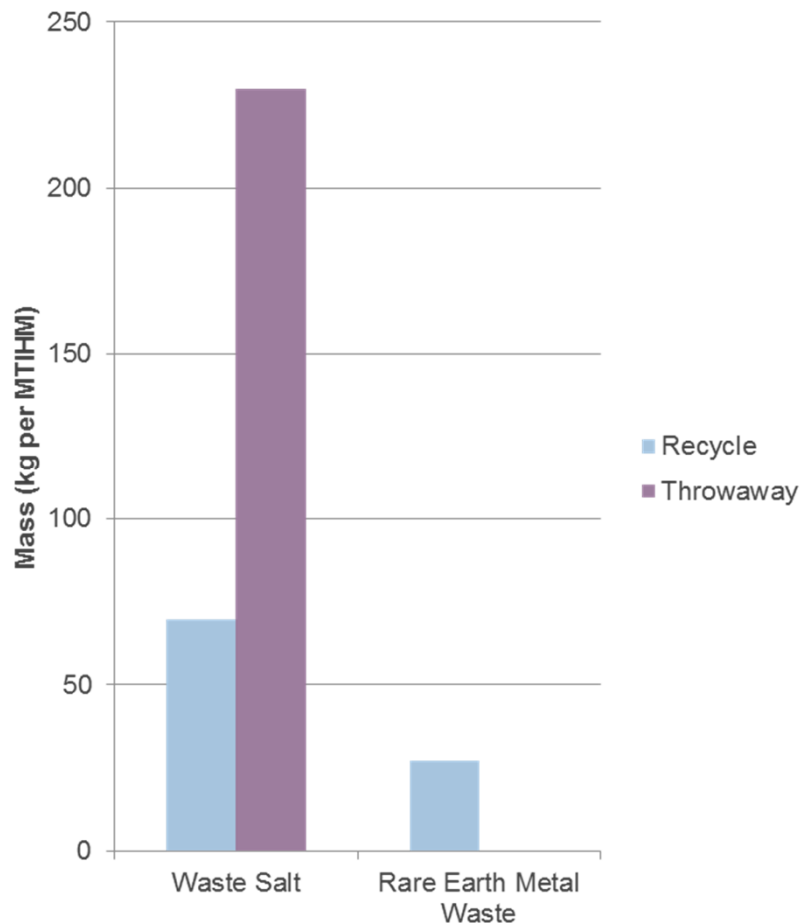


# Using AMPYRE to Evaluate Flowsheet Options (Oxide Fuel)

- Comparison of salt handling options
  - Recycle loop to remove rare earth and active metal fission products
  - Direct throwaway of electrorefiner salt following recovery of U/TRU
- Impact on waste



# Using AMPYRE to Evaluate Flowsheet Options (Metal Fuel)



- Throwaway salt after U/TRU recovery vs. recycle after removal of rare earth and active metals
- Assumptions:
  - Max 10 wt% NaCl
  - Recycle process removes 33% NaCl and 80% CsCl and SrCl<sub>2</sub>
- Salt treatment process needs to be defined



# Summary of FY-2014 Modeling Activities and Future Work

- Simulated system response to changes and variations
- Developed chemistry-based models for anode and cathode subunits
- Enhanced user interface to automate tasks and provide flexibility
- Integrate dynamic electrorefiner model with mass balance facility model
- Expand dynamic subunit models
- Link in-process observables with chemical process models
  - Provide inputs to monitors and sensors
  - Improve identification of dynamic trends
  - Support development of other codes





# Acknowledgements

- Government License Notice
  - The submitted manuscript has been created by UChicago Argonne, LLC, Operator of Argonne National Laboratory (“Argonne”). Argonne, a U.S. Department of Energy Office of Science laboratory, is operated under Contract No. DE-AC02-06CH11357. The U.S. Government retains for itself, and others acting on its behalf, a paid-up nonexclusive, irrevocable worldwide license in said article to reproduce, prepare derivative works, distribute copies to the public, and perform publicly and display publicly, by or on behalf of the Government.
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